
Condition That Exists When A Solution Contains More Solute Than Usual At Given Temperature

Condition That Exists When A Solution Contains More Solute Than Usual At Given Temperature - [FREE] **CONDITION THAT EXISTS WHEN A SOLUTION CONTAINS MORE SOLUTE THAN USUAL AT GIVEN TEMPERATURE** [EPUB] [PDF] The condition that exists when a solution cannot dissolve any more solute at a given temperature. ... The condition that exists when a solution contains more solute than usual at a given temperature. Temperature and Pressure-Typically, the solubility of a solid increases with an increase in the temperature of the solution.-The solubility of a ... - Tue, 13 Nov 2018 18:16:00 GMT **Chemistry Chapter 16 Study Guide Flashcards | Quizlet Quia - Chapter 8: Solutions, Acids, and Bases** Chapter 8: Solutions, Acids, and Bases. Solubility The maximum amount of solute that normally dissolves in a given amount of solvent at a certain temperature. **Contains more solute than a saturated solution does at a ...** A solution containing more solute than can theoretically be dissolved at a given temperature is called a super saturated solution. **Quia - Chapter 16 "Solutions"** At a given temperature, the solubility of a gas in a liquid is directly proportional to the pressure of the gas above the liquid: supersaturated solution: contains more solute than can theoretically be held at a given temperature: dilute solution: contains only a small amount of the maximum amount of solute that can be dissolved at a given ... **A solution which contains maximum solute at a given ...** klondikegj and 10 more users found this answer helpful I believe the solution that contains the maximum amount of solute present within it at a given temperature, would be referred to as saturated. No more solute can dissolve or be mixed within the solvent and have it dissolved at that temperature. **A solution which contains maximum solute at a given ...** Saturated solution contains the maximum concentration of a solute dissolved in the solvent (usually water) and if extra solute is add to saturated solution, that solute will not dissolve. The amount of solute that can be dissolved in a solvent depends of chemical composition, temperature and pressure. **What is a solution that contains more solute than it would ...** "Supersaturated....." "Supersaturation" is a metastable condition, and is rigidly defined as a solution that contains MORE solute than would be in equilibrium with undissolved solute at a given temperature.

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